

PS2 Solutions

1. The states are labeled by the number of open links s , where
 $\Sigma_s = S \cdot \varepsilon$ and $\Omega_s = 1$

$$Z = \sum_s \Omega_s e^{-\varepsilon_s/kT} = \sum_{s=0}^N e^{-s\varepsilon/kT} = \frac{1 - e^{-(N+1)\varepsilon/kT}}{1 - e^{-\varepsilon/kT}}$$

$$\langle s \rangle = \text{avg. \# of open links} = \frac{1}{\varepsilon} \langle E \rangle = \frac{1}{\varepsilon} kT^2 \frac{\partial}{\partial T} \log Z$$

= mess, in general

For $\varepsilon \gg kT$ $Z \approx 1 + e^{-\varepsilon/kT}$

$$\langle s \rangle = \frac{1}{\varepsilon} \cdot \frac{\varepsilon e^{-\varepsilon/kT}}{1 + e^{-\varepsilon/kT}} = \frac{1}{e^{\varepsilon/kT} + 1} < 1$$

why? it costs too much energy to hold a link open

For $\varepsilon \ll kT$ $Z \approx \frac{\frac{N+1}{kT} \varepsilon - \frac{1}{2} \left(\frac{N+1}{kT} \varepsilon \right)^2 + \dots}{\frac{\varepsilon}{kT} - \frac{1}{2} \left(\frac{\varepsilon}{kT} \right)^2 + \dots}$

$$= (N+1) \frac{1 - \frac{1}{2} \frac{N+1}{kT} \varepsilon + \dots}{1 - \frac{1}{2} \frac{\varepsilon}{kT} + \dots}$$

$$\approx (N+1) \left(1 - \frac{N\varepsilon}{2kT} + \dots \right)$$

$$\log Z = \log(N+1) - \frac{N\varepsilon}{2kT}$$

$$\langle s \rangle = \frac{1}{\varepsilon} \cdot kT^2 \cdot \frac{N\varepsilon}{2kT^2} = \frac{N}{2} \quad \text{! open + closed states are equally likely}$$

$$2. \quad S = \frac{E - F}{T} \quad \text{from the thermo. definition of } F$$

$$= \frac{1}{T} \sum_s \varepsilon_s p_s + k_B \log Z$$

$$= -k_B \sum_s p_s \left(\underbrace{\frac{-\varepsilon_s}{k_B T}}_{= \log p_s} + \log Z \right) \quad \text{since } \sum_s p_s = 1$$

$$= -k_B \sum_s p_s \log p_s$$

This formula also works for the micro-canonical ensemble
where $p_s = \frac{1}{\Omega(E)}$

$$-k_B \sum_s p_s \log p_s \rightarrow k_B \log \Omega(E) \sum_s p_s = k_B \log \Omega(E).$$

$$2. \quad S = \lambda V^{1/2} (NE)^{1/4}$$

$$(a) \quad T^{-1} = \left(\frac{\partial S}{\partial E} \right)_{V, N} = + \frac{1}{4} \lambda V^{1/2} N^{1/4} E^{-3/4} = S/4E$$

$$p = T \left(\frac{\partial S}{\partial V} \right)_{E, N} = T \cdot \frac{1}{2} \lambda V^{-1/2} (NE)^{1/4} = TS/2V$$

$$\hookrightarrow \text{from } dE = TdS - pdV \rightarrow dS = \frac{1}{T}dE + \frac{p}{T}dV$$

(c) In equilibrium $T_1 = T_2$ so

$$V_1^{1/2} N_1^{1/4} E_1^{-3/4} = V_2^{1/2} N_2^{1/4} E_2^{-3/4}$$

$$\text{or } \left(\frac{E_2}{E_1} \right)^{3/4} = \left(\frac{V_2}{V_1} \right)^{1/2} \left(\frac{N_2}{N_1} \right)^{1/4} = 1 \cdot \left(\frac{2/3}{1/3} \right)^{1/4}$$

$$\Rightarrow E_2/E_1 = 2^{1/3}$$

(e) Here $T_1 = T_2$ and $p_1 = p_2$

$$T: \quad V_1^{1/2} \left(\frac{1}{3} \right)^{1/4} E_1^{-3/4} = V_2^{1/2} \left(\frac{2}{3} \right)^{1/4} E_2^{-3/4}$$

$$p: \quad V_1^{-1/2} \left(\frac{1}{3} \right)^{1/4} E_1^{1/4} = V_2^{-1/2} \left(\frac{2}{3} \right)^{1/4} E_2^{1/4}$$

$$\text{divide: } V_1/E_1 = V_2/E_2$$

$$\text{multiply: } \left(\frac{1}{3} \right)^{1/2} E_1^{-1/2} = \left(\frac{2}{3} \right)^{1/2} E_2^{-1/2}$$

$$\Rightarrow E_2/E_1 = 2$$